

R. Friary* and B. R. Sunday

Research Division, Schering-Plough Corporation, Bloomfield, N. J. 07003

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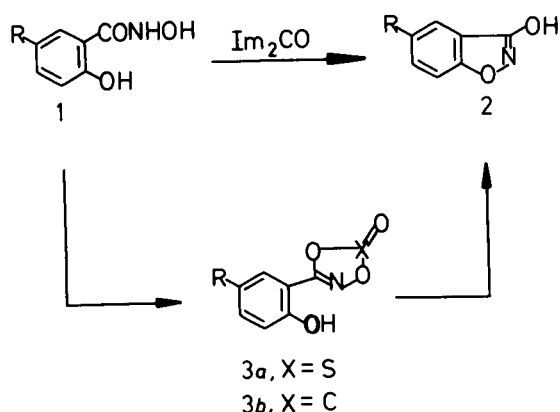
Carbonyldiimidazole cyclizes 2-hydroxybenzohydroxamic acids to 3-hydroxy-1,2-benzisoxazoles.

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Cyclodehydration of 2-hydroxybenzohydroxamic acids (**1**) gives 3-hydroxybenzisoxazoles (**2**) in two stages (1,2). Under necessarily acidic conditions, thionyl and carbonyl chlorides change **1** to **3a** and **3b**, respectively, and bases like triethylamine and pyridine then convert **3a-b** to **2** (1,2).

These facts imply that **2** might be prepared directly if **1** could be changed to **3b** under basic conditions. Were any substituent of **1** oxidizable, they suggest that another reagent might yield more of **3** -- and hence of **2** -- than thionyl chloride.

Indeed, **3b** (R = CH₃CO) was formed when **1c** (R = CH₃CO) was treated at room temperature with 1.1 equivalents of carbonyldiimidazole, and brief treatment of **1a-b** with 2 equivalents of this reagent in hot tetrahydrofuran followed by acidification gave the corresponding, known 3-hydroxybenzisoxazoles **2a** (R = Cl, 68%) and **2b** (R = OMe, 42%). Although **2c** (R = CH₃CO) could not be prepared with thionyl chloride and triethylamine, carbonyldiimidazole and **1c** similarly yielded 95% of **2c**.



EXPERIMENTAL (3)

5-Chloro-2-hydroxybenzohydroxamic Acid (**1a**).

This compound, m.p. 231-232° (lit. m.p. 325° (1), m.p. 246° (4)), was prepared according to Böhshagen (1).

2-Hydroxy-5-methoxybenzohydroxamic Acid (**1b**).

This compound, m.p. 172-175° (lit. m.p. 185° (1)), was prepared according to Böhshagen (1).

Methyl 5-Acetyl-2-hydroxybenzoate.

This compound, m.p. 55-60° (lit. m.p. 60-62° (5), m.p.

55° (6)), was prepared in a yield of 81% from methyl 2-hydroxybenzoate, aluminum chloride and acetyl chloride by a Friedel-Crafts reaction in tetrachloroethane.

Methyl 5-(1,1-Dimethoxy)-ethyl-2-hydroxybenzoate.

A solution of methyl 5-acetyl-2-hydroxybenzoate (50.2 g., 0.258 mole), trimethylorthoformate (42.2 ml.), *p*-toluenesulfonic acid monohydrate (100 mg.), and methanol (250 ml.) was boiled under reflux for 2 hours; it was cooled, and sodium bicarbonate (500 mg.) was added. The solvents were evaporated and the residue in ether was washed with 1M sodium bicarbonate solution, water, and brine. The solution was dried over sodium sulfate, filtered and evaporated. Distillation of the residue gave 54.6 g. (88%) of the desired product, b.p. 96-104° at 0.15 mm. Crystallization from hexanes gave the analytical sample, m.p. 40-41°.

Anal. Calcd. for C₁₂H₁₆O₅: C, 59.99; H, 6.71%. Found: C, 60.03; H, 6.70%.

5-Acetyl-2-hydroxybenzohydroxamic Acid (**1c**).

A. From Methyl 5-(1,1-Dimethoxy)-ethyl-2-hydroxybenzoate.

According to the method of Böhshagen (1), this compound (0.23 mole) gave 93% of **1c**, identified by comparison of its ir spectrum to that of an authentic sample; deketalization occurred during work-up with aqueous hydrochloric acid.

B. From 2-Hydroxybenzohydroxamic Acid.

Acetyl chloride (21.4 g., 0.329 mole) was added dropwise to a stirred solution of aluminum chloride (57.5 g., 0.432 mole) in nitrobenzene (200 ml.) at 25°; the mixture was then stirred at 25° for 0.5 hour. In five equal portions, a total of 25.0 g. (0.163 mole) of 2-hydroxybenzohydroxamic acid was added during 1 hour; the internal temperature was kept at 20 to 30° by ice cooling. After 3 hours at 20°, the solution was added dropwise to ice water (1.6 kg.); the precipitate (29.6 g., m.p. 157-163° dec.) was collected, dried and crystallized from acetic acid to give 8.10 g. (25.5%) of the desired product, m.p. 213-217° dec. as a pink solid. Recrystallization from methanol gave the analytical sample.

Anal. Calcd. for C₉H₉NO₄: C, 55.39; H, 4.65; N, 7.18%. Found: C, 55.60; H, 4.75; N, 7.20%.

5-Acetyl-3-hydroxy-1,2-benzisoxazole (**2c**).

A solution of carbonyl diimidazole (62.0 g., 0.380 mole) in tetrahydrofuran (550 ml.) was added to a boiling solution of **1c** (37.0 g., 0.190 mole) in tetrahydrofuran (385 ml.); the resulting solution was then boiled under reflux for 1 hour and was cooled and evaporated. The residue was dissolved in water (330 ml.), and the solution was cooled and acidified to pH 2 with concentrated hydrochloric acid (100 ml.). The precipitate was collected and crystallized from ethyl acetate (using a Soxhlet extractor overnight) to give 30.0 g. (89%) of the desired

product, m.p. 217.0-219.5°, as white needles. Two recrystallizations of the residue gave an additional 1.92 g., m.p. 212-216°, raising the yield to 95%; ¹³C-nmr (DMSO-*d*₆): (25 MHz) δ C₃, 167.0; C_{3a}, 115.0; C₄, 123.3*; C₅, 132.6; C₆, 130.3*; C₇, 110.3; C_{7a}, 165.3; CO, 190.0; CH₃, 26.7 ppm

Anal. Calcd. for C₉H₇NO₃: C, 61.02; H, 3.98; N, 7.91. Found: C, 61.31; H, 3.92; N, 7.88.

3-(5-Acetyl-2-hydroxyphenyl)-1,4,2-dioxazolin-2-en-5-one (**3b**, R = CH₃CO).

When **1c** was treated at room temperature with only 1.1 equivalents of carbonyldiimidazole in tetrahydrofuran, a small amount of **3b** (R = CH₃CO), m.p. 186-188°, was isolated instead of **2c**.

Anal. Calcd. for C₁₀H₇NO₅: C, 54.31; H, 3.19; N, 6.33. Found: C, 54.14; H, 2.97; N, 6.30.

Attempted Conversion of **1c** to **2c** with Thionyl Chloride.

In each of three trials, sequential treatment of **1c** with thionyl chloride and triethylamine according to Böshagen's method (1) gave an intractable mixture.

5-Chloro-3-hydroxy-1,2-benzisoxazole (**2a**).

Treatment of **1a** with two equivalents of carbonyldiimidazole in boiling tetrahydrofuran as described for **2c** gave, after crystallization from methanol, 68% of **2a**, m.p. 216-220° (lit. m.p. 220° (1)); ¹³C-nmr (DMSO-*d*₆): (25 MHz) δ C₃, 165.0; C_{3a}, 116.3; C₄, 120.8*; C₅, 127.5; C₆, 130.6*; C₇, 111.8; C_{7a}, 162.0 ppm.

Anal. Calcd. for C₇H₄ClNO₂: C, 49.58; H, 2.38; Cl, 20.91; N, 8.26. Found: C, 49.55; H, 2.20; Cl, 20.88; N, 8.38.

3-Hydroxy-5-methoxy-1,2-benzisoxazole (**2b**).

Treatment of **1b** with two equivalents of carbonyldiimidazole in boiling tetrahydrofuran as described for **2c** gave, after crystallization from methanol, 42% of **2b**, m.p. 175-177° (lit. m.p. 183° (1)).

Anal. Calcd. for C₈H₇NO₃: C, 58.18; H, 4.27; N, 8.48. Found: C, 57.90; H, 4.01; N, 8.52.

Acknowledgement.

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